The use of infra-re

The use of infra-red absorption and molar refractivities to check coordination

By Frank Dachille and Rustum Roy College of Mineral Industries, The Pennsylvania State University University Park, Pennsylvania*

With 3 figures

(Received June 6, 1959)

Auszug

Durch experimentelle Untersuchung der Ultrarot-Absorptionsspektren einiger SiO₂-Isotope war es möglich, die primäre Koordination, die Atomordnungszahl, das Atomgewicht und die Ionenladung den Absorptions-Wellenlängen der größeren Banden zuzuordnen. Auch die Verwendung der Molrefraktion zur Voraussage und Bestimmung von Änderungen der (primären und sekundären) Koordination wird erläutert.

Abstract

By an empirical study of the infra-red absorption spectra of some silica isotypes it has been possible to relate primary coordination, atomic number, atomic weight and ionic charge to absorption wave lengths for the major bands. Likewise the utilization of molar refractivity in predicting or determining coordination changes (both primary and secondary) is illustrated.

Introduction

An attempt has been made at using infra-red absorption spectra and molar refractivities to yield information on coordination of cations in simple structures. The fact that infra-red absorption spectra will distinguish phases, which are quite similar crystallographically, is already well known¹, and there is no doubt that the differences in such spectra reflect even minor changes of bond length and configuration. Of course, even in slightly complex uniaxial structures it is difficult to make an intelligent guess at assigning particular frequencies to certain vibration modes.

To proceed from the spation modes is straightforware even to detect all expected largely, no doubt, to the vecases, such as quartz itself possible to group the absocation-anion stretching, catanion bending and distort stretching occurs at the strongest absorption. In a cation-anion bond is the cations at highest frequency straightforward even to detect all expected largely, no doubt, to the vecases, such as quartz itself possible to group the absocation-anion stretching, cat anion bending and distort stretching occurs at the straightforward even to detect all expected largely, no doubt, to the vecases, such as quartz itself possible to group the absocation-anion stretching, cat anion bending and distort stretching occurs at the strongest absorption. In a cation-anion bond is the cation-anion bending and distort stretching occurs at the strongest absorption.

Moreover, in simple stru hope that certain associatio of known-structure materi groups of phases for such a groups of polymorphs.

These same phases, es value in studying changes i are available which exist ir forms, each of which can substance is GeO₂ which ex four-coordinated quartz for

Infra-

In our early work² on show clearly the ability to absorption spectroscopy. I the information and deduct nature of these results is of from their usefulness.

The most significant sp morphs. Spectra for the poly SiO₂ cristobalite, SiO₂ glass isomorphous series BPO₄,

^{*} Contribution No. 58-123.

¹ Rustum Roy and E. E. Francis, On the distinction of sillimanite from mullite by infra-red techniques. Am. Min. 38 (1953) 725—728.

² Rustum Roy, Isomorphia Symposium on "Defect structu